

EXHIBIT D

Polyethylene glycol-conjugated pharmaceutical proteins

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Polyethylene glycol (PEG)-conjugated proteins belong to a new class of biomolecules that are neither proteins nor polymers, but which are hybrids of the two. PEG conjugation of biopharmaceuticals is now common practice in efforts to achieve sustained clinical response. However, very little is known about the strategies and criteria used to produce a well-characterized pegylated biomolecule. In this review, the issues of pegylation reagent selection, reaction conditions, process considerations and purification will be addressed, as well as biochemical and biological characterizations. The isolation of positional isomers and the determination of pegylation sites will also be included.

(Ref. 32), CD4-IgG (Ref. 33) and interferon- γ (IFN- γ) (Ref. 34), among others.

Rationale

In addition to the aforementioned beneficial properties of pegylated proteins, pegylation can impart the following advantages:

- altered biodistribution;
- altered biological properties²⁵⁻²⁷ (see Table 1);
- enhanced membrane penetration³¹, sustained clinical response with minimal dosing leading to improved quality of life and reduced treatment cost.

Conceptual considerations

In general, PEG-conjugated biomolecules exhibit physicochemical properties that are different from those of the parent molecules. These properties include conformational changes, steric interference, changes in electrostatic-binding properties, hydrophobicity, local lysine pKas (a measurement of the degree of completeness of a reversible reaction) and pI (the pH at which a protein's charge is neutral). Binding affinities to the receptors are often affected by these physicochemical changes, resulting in reduced activities in cell-based assays, in which incubation times are usually short. There is a direct relationship between the attached PEG mass and the *in vivo* biological activity¹⁹. In contrast, an inverse relationship exists between the PEG mass and the *in vitro* activity²². These relationships are illustrated in Fig. 1. Pegylation results in the retardation of renal clearance, which in turn prolongs the circulating half-life of the molecule and increases the area under the curve. The increased biological activity is attributed to these phenomena. In cell-based *in vitro* assays, because incubation is performed for a fixed time period, the enhanced half-life of the pegylated biomolecule has no role

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♥ Recombinant proteins are noted for their short circulating half-life in blood. This shortcoming is accompanied by reduced bioavailability and consequently, reduced clinical potency. In 1977, Abuchowski and colleagues demonstrated that, as therapeutic agents, polyethylene glycol (PEG)-conjugated proteins are more effective than their corresponding unmodified parent molecules¹. Since then, several pharmaceutical proteins have been pegylated and have been shown to have properties of use in clinical applications²⁻⁵. These improved clinical properties include better physical and thermal stability⁶⁻⁸, protection against susceptibility to enzymatic degradation⁷⁻¹¹, increased solubility¹²⁻¹⁴, longer *in vivo* circulating half-life¹²⁻¹⁴, decreased clearance and enhanced potency. Three other properties of pegylated proteins are reduced immunogenicity^{1,22-28} and antigenicity^{1,22,26,29,30}, as well as reduced toxicity¹⁹⁻²⁸. A further effect of pegylation is reduced *in vivo* activity accompanied by enhanced *in vitro* activity. This effect has been observed with granulocyte-macrophage colony stimulating factor (GM-CSF) (Ref. 26), interleukin-2 (IL-2) (Ref. 14), tumor necrosis factor- α (TNF- α) (Ref. 31), IL-6

Table 1. Altered biological properties of pegylated biopharmaceuticals

PEG-proteins	Altered bioactivity	Reference
GM-CSF	Neutrophil priming ^a Colony stimulating ^a	Knusi <i>et al.</i> ³⁶
IFN α -2a	Antiviral ^a Antiproliferative ^a	Falleron <i>et al.</i> (pers. commun.)
IL-15	CTL proliferation ^a Antagonistic ^a	Pettit <i>et al.</i>
Cholesterol esterase	HDL cholesterol ^a LDL cholesterol ^a	Suguichi <i>et al.</i> ³⁷
Cholesterol	HDL cholesterol ^a LDL cholesterol ^a	Suguichi <i>et al.</i> ³⁸

^aIncreased bioactivity.

^bDecreased bioactivity.

Abbreviations: PEG, polyethylene glycol; GM-CSF, granulocyte-macrophage colony stimulating factor; IFN, interferon; IL, interleukin; CTL, cytotoxic lymphocyte leukemia (a murine cell line); HDL, high density lipoprotein; LDL, low density lipoprotein.

to play. The relatively short incubation time, combined with the low receptor binding affinity of pegylated proteins, cause a lowering of *in vitro* bioactivity.

PEG attachment: unisite vs multiple sites

It is estimated that an optimal PEG mass of 40–60 kDa is required to retard renal and cellular clearance^{33,39}. The attachment of PEG to protein can be achieved in three different ways: a single large PEG at a single site, a branched PEG (two- or more medium PEG chains joined together via a linker) at a single site, or several small chains at multiple sites. In theory, unisite pegylated proteins should have higher activity because of a reduced chance of PEG attachment occurring at receptor-

binding domains. Multiple sites of pegylation or the attachment of large PEGs may result in partial- or complete loss of bioactivity.

Overlapping bioactivities

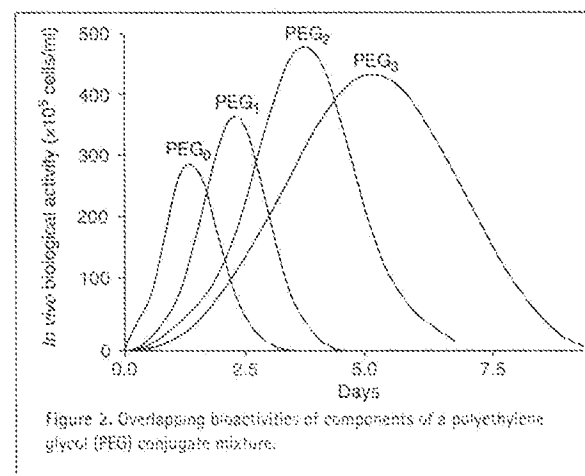
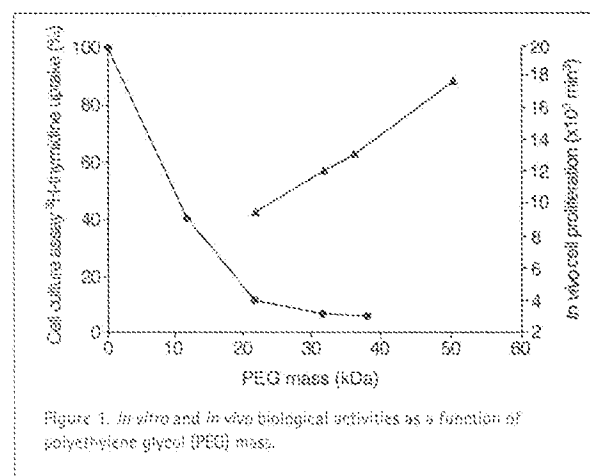
Potentially, there is an advantage in having a PEG-protein conjugate mixture comprised of unmodified, mono-, di-, tri-pegylated and other higher forms of protein PEG-conjugates. In theory, the higher the degree of pegylation, the slower the rate of absorption (prolonging the duration of circulation) and receptor saturation. The overlapping activity expected from such a mixture because of various absorption rates is illustrated in Fig. 2.

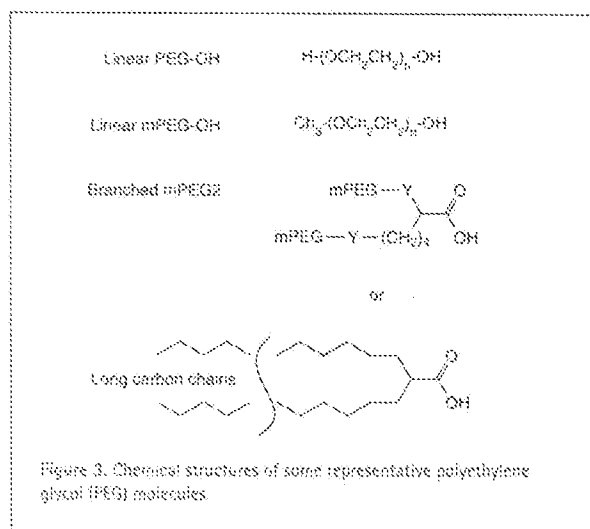
Polyethylene glycols (PEG)

Polyethylene glycols are amphiphilic polymers comprised of repeating ethylene oxide subunits, whose number is represented by the whole integer *n* (Fig. 3). Each ethylene oxide residue has a molecular weight (MW) of 44 Da, and $n \times 44$ Da represents the number average MW of the PEG chain. Polyethylene glycols are inert, nontoxic and contain two terminal hydroxyl groups that can be chemically activated. However, it is common practice to convert one of the two hydroxyl groups to methoxy or other alkoxy groups, in order to make the PEG unfunctional. In addition to linear PEG chains, there are also branched PEGs in which two or more PEG chains are joined together with linkers such as lysine⁴¹ and triazine⁴², among others. Figure 3 illustrates some representative PEG molecules.

Pegylation chemistries

A variety of pegylation chemistries and reagents are now readily available⁴³. They differ in their relative chemical reactivity





and specificity. The most commonly used pegylation reaction involves an electrophilically activated PEG and the ϵ -amino group of lysine or the protein's N-terminal amino group. Examples include:

- N-Hydroxysuccinimide-activated esters^{23,24,25,26} (amide bond);
- PEG-Epoxyde²⁶ (alkyl bond);
- PEG-Carbonyl imidazole²⁷ (urethane bond);
- PEG-2-Nitrophenyl carbonates²⁸ (urethane bond);
- PEG-Triethylate²⁹ (alkyl bond);
- PEG-Aldehyde³⁰ (N-terminus, Schiff's base).

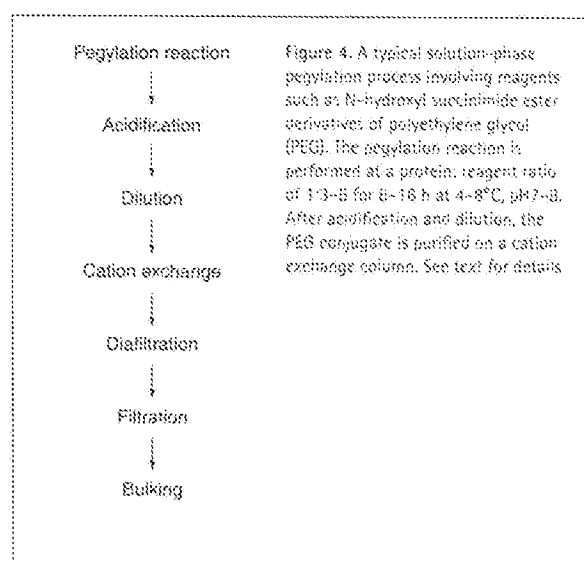
In addition to the attachment of PEG via primary amino groups which often produces heterogeneous PEG conjugates, site-directed pegylation of proteins can be achieved via other functional groups on the protein surface. These include, free cysteines, oligosaccharides and alcoholic groups. PEG reagents used for site-specific pegylations are:

- PEG-Vinyl sulphone³¹ (via free cysteine);
- PEG-Iodoacetamide³² (via free cysteine);
- PEG-Maleimide³³ (via free cysteine);
- PEG-orthopyridyl disulfide³⁴ (via free cysteine);
- PEG-Hydrazide³⁵ (via oligosaccharide);
- PEG-Isocyanate³⁶ (via alcohol or amino group).

Pegylation reaction parameters

Factors that affect pegylation reactions are:

- reaction pH;
- protein to PEG-reagent molar ratio;



- protein concentration;
- time of reaction;
- reaction temperature.

By controlling one or more of these factors, the reaction can be directed towards producing predominantly mono-, di-, tri-, etc., PEG conjugates.

Pegylation process considerations

The pegylation of proteins can be performed in solution-phase or in a continuous solid-phase mode

Solution phase

In a typical solution phase pegylation reaction involving PEG reagents such as N-hydroxysuccinimide esters, the final protein-to-reagent molar ratio used is usually 1:3-5. The reaction pH is maintained at around 7-8, the temperature is kept at 4-8°C and the reaction is terminated after 8-16 h by adjusting the pH to 4.5 with glacial acetic acid. The reaction mixture is then diluted with water and adsorbed onto a cation exchange column. Excess reagent and reaction byproducts are washed away. Pegylated protein is then separated from the unmodified protein by using increasing concentrations of salt in the buffer. A typical purification scheme is illustrated in Fig. 4.

Solid-phase pegylation

In solid-phase pegylation³⁷, the protein to be pegylated is adsorbed onto an anion exchange column and the PEG reagent is circulated through the column for a predetermined period of time. Excess reagent and other reaction byproducts are removed from the column by washing with buffer. The salt concentration

of the elution buffer is chosen so that only the pegylated protein elutes, while the unmodified protein remains on the column. From the eluate, the amount of protein pegylated is determined and the column is replenished with an equivalent amount of unmodified protein, so that the original amount of protein is maintained at the start of the next cycle of operation. PEG reagent is circulated again and the modified protein is separated from the unmodified as in the first cycle of operation. In the solid-phase pegylation process, reaction, separation and purification are performed on the same column and can be repeated as many times as required.

Biochemical and biological characterization

Protein determination

Protein concentration is determined by standard protein assay methods such as, UV absorbance, Lowry, bicinchoninic acid (BCA) (used in protein assays for color development), Bradford, etc., or by the more reliable and accurate amino acid composition-analysis method.

SDS-PAGE analysis

SDS-PAGE is performed according to the methods of Laemmli⁵⁸, and the gels can be stained specifically for protein or PEG. Staining for PEG is done according to a modified procedure of Kurfurst⁵⁹. SDS-PAGE gels are rinsed with distilled water and placed in a 5% barium chloride solution for ten minutes. After rinsing the gel with distilled water to remove barium chloride, the gel is placed in 0.1 M Trisol™ (EM Science, Gibbstown, NJ, USA) for another ten minutes. After washing away the yellow Thimol™ (iodine solution), the gel is stored in distilled water in a heat-sealed Kapak/Scotchpak bag in the dark.

Molecular weight determination and identification

Mass spectrometry. Though not quantitative, mass spectrometry is an ideal tool in determining the true MW of various PEG conjugate species. Matrix-assisted laser desorption time of flight mass spectrometry (MALDI TOF MS) is now an established method for not only determining the true MW of PEG conjugate species, but also in identifying the individual species contained in a particular preparation^{60,61}.

SDS-PAGE analysis. Because of the anomalous (retarded) electrophoretic mobility of PEG moieties, SDS-PAGE analysis gives higher than expected MW, when compared with MW marker proteins. In order to circumvent this problem one can use PEG MW standards, and stain specifically for PEG.

Determination of pegylation sites and positional isomers⁶⁰⁻⁶³. Pegylation sites and the corresponding positional isomers are determined using a combination of techniques. These techniques include,

ion exchange- HPLC, size-exclusion chromatography, reversed-phase HPLC, peptide mapping, Edman sequencing, amino acid analysis and MALDI TOF MS. The potential number of positional isomers in a pegylated molecule can be calculated using the following factorial equation⁶²:

For *N* pegylation sites, taken *k* at a time,

$$\frac{N!}{(N - K)! \times K!} = \text{possible number of combinations}$$

Bioactivities

As illustrated in Fig. 1, there is an enhancement in *in vivo* biological activity relative to the PEG mass attached, whereas a decrease in *in vitro* activity is observed with an increase in PEG-mass attachment⁶⁴. It should be pointed out that in almost all cases, the *in vitro* activity is not predictive of the *in vivo* activity. For reasons explained in the section on conceptual considerations, *in vivo* activity is many fold higher than that observed in an *in vitro* assay.

Pharmacokinetics

When injected into animals, pegylated biomolecules exhibit superior pharmacokinetic properties than those exhibited by the unmodified parent molecule. The biological activity is long-lasting because the serum circulating half-life and the plasma residence time of a PEG conjugate are often many fold higher than those of the unmodified molecule¹⁷⁻²².

Immunogenicity

In immunogenicity a much reduced antibody response is observed with pegylated proteins when compared with those observed with an unmodified- or aggregated form of the unmodified protein^{17,23,24}.

Toxicity

In terms of toxicity, overall pegylated proteins are less toxic than the unmodified parent molecule^{19,25}. However, toxicities have been observed with topically applied low MW PEGs⁶⁷. These adverse effects are not observed however, when low dosages of PEG are administered orally or parenterally⁶². Because only chronic administration of large quantities of PEG elicits such toxicities, it is not expected to be the case with pegylated proteins such as cytokines, enzymes and hormones.

Tissue uptake of polyethylene glycol

Small PEG's tend to translocate freely from circulation into extravascular tissues and diffuse back into circulation⁶⁸. However, a much slower diffusion process is observed for large PEG molecules. The larger the size of the PEG, the lower the renal clearance. In contrast, liver clearance increased with

higher MW PEGs. When the MW of the PEG is >50 kDa, the uptake by Kupfer cells is enhanced.

Regulatory guidelines of pegylated biopharmaceuticals

When developing pegylated biomolecules as therapeutic agents, the following points should be considered:

- starting materials (protein and PEG reagent) should be well-characterized;
- pegylated proteins are a new class of molecules;
- pegylated proteins are heterogeneous: neither proteins nor polymers, but hybrids of the two;
- sites of pegylation and stoichiometry of PEG attachment should be determined;
- consistency of each preparation should be established;
- derivatization and manufacturing processes should be validated.

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References

1. Abuchowski, A. et al. (1977) *J. Biol. Chem.* **252**, 3578-3581
2. Delgado, C., Francis, G.K. and Fisher, D. (1992) *Cell. Res. Ther. Drug Cancer Syst.* **9**, 249-264
3. Eare, N.V. (1993) *Adv. Drug Deliv. Syst.* **10**, 91-114
4. Zalipsky, S. (1995) *Bioconj. Chem.* **6**, 150-167
5. Inada, Y. (1990) *J. Bioact. Compat. Polym.* **5**, 343-363
6. Sozúto, T. et al. (1984) *Biochem. Biophys. Acta* **788**, 246-255
7. Hadley, E.B. and Sato, H.H. (1989) *Enzyme* **42**, 223-234
8. Hallington, M.W. and Sommer, P.E. (1988) *Ann. New York Acad. Sci.* **542**, 244-249
9. Cao, S.C. et al. (1990) *Ann. New York Acad. Sci.* **613**, 440-467
10. Abuchowski, A. et al. (1977) *J. Biol. Chem.* **252**, 3582-3586
11. Liu, P.L. et al. (1991) *J. Appl. Biochem.* **4**, 17-27
12. Ashihara, Y. et al. (1978) *Biochem. Biophys. Res. Commun.* **83**, 385-393
13. Abuchowski, A. and Davies, S.A. (1981) in *Bioproc. as Drugs* (Goldenberg, J.C., ed.), pp. 367-377. John Wiley & Sons, New York, NY, USA
14. Eare, N.V., Enaul, M.J. and Laird, W.J. (1987) *Proc. Natl. Acad. Sci. U.S.A.* **84**, 1487-1491
15. Chen, B.H. et al. (1981) *Biochim. Biophys. Acta* **660**, 293-298
16. Davis, P.F. et al. (1980) in *Biomedical Polymers, Polymer Materials and Pharmaceuticals for Biomedical Use* (Goldberg, H.P. and Nakajima, A., eds.), pp. 441-452. Academic Press, New York, NY, USA
17. Enaul, M.J. et al. (1988) *J. Biol. Chem.* **263**, 15064-15070
18. Hu, D.H. et al. (1986) *Drug Mark. Dupes.* **14**, 349-352
19. Paerreges, P. and Abuchowski, A. (1990) *J. Control. Release* **11**, 139-148
20. Enake-Ishtikova, R. (1992) *Cell Struct. Funct.* **17**, 157-160
21. Horami, H. (1986) *US Patent* **4,602,546**
22. Most, A.H. et al. (1990) *Biol. Chem. Hoppe-Seyler* **371**, 101-109
23. Eare, N. (1990) *J. Immunol.* **144**, 209-213

24. Abuchowski, A. (1984) *Cover Biochem. Biophys.* **7**, 175-186
25. Kamisaki, S. et al. (1981) *J. Pharmacol. Exp. Ther.* **216**, 410-413
26. Davis, S. et al. (1981) *Clin. Exp. Immunol.* **46**, 649-652
27. Nucci, M.L., Ojapany, K. I. and Abuchowski, A. (1986) *J. For. Biol. Med.* **2**, 311-325
28. Tang, L. et al. (1981) *Int. J. Immunopharmacol.* **7**, 715-730
29. Yokoyama, M. et al. (1990) *Cancer Res.* **50**, 1693-1706
30. Veronese, G.M. et al. (1985) *Appl. Biochem. Biotechnol.* **11**, 141-151
31. Tatemura, T. et al. (1994) *Jpn. J. Cancer Res.* **85**, 9-12
32. Giron, H. et al. (1994) *Lab. Clin. Med.* **124**, 528-536
33. Charnow, S.M. et al. (1994) *Bioconj. Chem.* **5**, 133-140
34. Kita, Y. et al. (1990) *Drug Del. Deliv.* **6**, 157-167
35. Kozul, C. et al. (1992) *Br. J. Haematol.* **654**-663
36. Pettit, D.K. et al. (1997) *J. Biol. Chem.* **272**, 2312-2318
37. Suguchi, H. et al. (1995) *Clin. Chem.* **41**, 717-723
38. Beckman, J.S. et al. (1988) *J. Biol. Chem.* **263**, 6884-6892
39. Fung, W.J., Porter, J. and Hailon, P. (1997) *Polym. Prepr.* **38**, 565-566
40. Yamazaki, Y., Takata, Y. and Ikada, Y. (1994) *J. Pharm. Sci.* **83**, 691-696
41. Monfardini, C.S. et al. (1985) *Bioconj. Chem.* **6**, 62-69
42. Harris, J.M. and Zalipsky, S. eds. (1997) in *Poly(ethylene Glycol) Chemistry and Biological Applications* (Harris, J.M. and Zalipsky, S. eds.), pp. 1-474. American Chemical Society, Washington, DC, USA
43. Kabanov, A.L. (1990) *PER Lett.* **268**, 235-237
44. Zalipsky, S. and Lee, C. (1992) in *Biomedical Applications of Polyethylene Glycol Chemistry* (Harris, J.M., ed.), pp. 347-370. Plenum Press, New York, NY, USA
45. Zalipsky, S. et al. (1987) *Int. J. Pept. Protein Res.* **30**, 740-753
46. Elling, L. and Nola, M.R. (1991) *Biochem. Appl. Biochem.* **13**, 354-362
47. Bouchamp, C.C. et al. (1983) *Anal. Biochem.* **131**, 25-33
48. Veronese, G.M. et al. (1985) *Appl. Biochem. Biotechnol.* **11**, 141-151
49. Delgado, C. et al. (1990) *Biochem. Appl. Biochem.* **12**, 119-128
50. Wirth, P. et al. (1991) *Imm. Chem.* **19**, 133-142
51. Morpurgo, M. et al. (1994) *J. Bioconj. Chem.* **7**, 363-368
52. Shearwater Polymers, Inc. (1998) *Quosely Plasmid*, pp. 1-4. Shearwater Polymers Inc., Huntsville, AL, USA
53. Crookson, R.J. and Eare, N.V. (1990) *Int. J. Lab. Med.* **8**, 343-346
54. Waghire, C., Sharma, B. and Srivast, S. (1993) *Bioconj. Chem.* **4**, 314-318
55. Zalipsky, S. and Menon-Rudolph, S. (1997) in *Poly(ethylene Glycol) Chemistry and Biological Applications* (Harris, J.M. and Zalipsky, S., eds.), pp. 318-341. American Chemical Society, Washington, DC, USA
56. Greenwald, R.B., Pandri, A. and Bolikal, D. (1992) *J. Org. Chem.* **60**, 311-338
57. Porter, J. et al. (1996) *Symposium on Biomedical Products VIII*, sponsored by Div. of Biochem. Technol., 20-25 October, Tucson, AZ, USA
58. Laemmli, U.K. (1970) *Nature* **227**, 680-685
59. Kufner, M. (1992) *Anal. Biochem.* **200**, 244-248
60. Monfardini, S.P. et al. (1997) in *Poly(ethylene Glycol) Chemistry and Biological Applications* (Harris, J.M. and Zalipsky, S., eds.), pp. 207-216. American Chemical Society, Washington, DC, USA
61. Monfardini, S.P. et al. (1997) *Anal. Biochem.* **242**, 419-440
62. Burns, D.H. et al. (1982) *Imm.* **9**, 49-52